ESR studies on iron-sulfur clusters of complex II in Ascaris suum mitochondria which exhibits strong fumarate reductase activity

Akiko Hata-Tanaka, Kiyoshi Kita⁺, Rieko Furushima⁺, Hiroshi Oya⁺ and Shigeru Itoh[†]

Department of Biology, Faculty of Science, University of Tokyo, Hongo, Tokyo 113, *Department of Parasitology, Juntendo University of Medicine, Hongo, Tokyo 113 and *National Institute for Basic Biology, Okazaki, Aichi 444, Japan

Received 22 September 1988; revised version received 24 October 1988

Complex II of Ascaris suum mitochondria, which functions as fumarate reductase in physiological conditions, contains three types of iron-sulfur clusters. These correspond to clusters S-1, S-2 and S-3 and are distinguishable by low-temperature ESR studies. Cluster S-1 is reduced by succinate, giving ESR signals with g_z , g_y and g_x values at 2.033, 1.939 and 1.920. The existence of cluster S-2 is suggested by an enhancement of the S-1 spin relaxation induced upon reduction of S-2 by dithionite. Cluster S-3 is ESR detectable under air-oxidized conditions and gives a strong signal at g = 2.025. Cluster S-3 was only partially reduced even with an excess amount of sodium succinate, which is a common characteristic of fumarate reductase but this is not seen in the mitochondrial complex II.

Iron-sulfur cluster; Complex II; Fumarate reductase; ESR; (Ascaris suum, Nematoda)

1. INTRODUCTION

Adult Ascaris suum resides in the host's small intestine, and anaerobically respires with electron transfer from NADH to fumarate [2]. Reducing equivalents from NADH are accepted by the mitochondrial complex I, mediated by rhodo-quinone, and oxidized by the complex II fumarate reductase [2-4]. Thus, the mitochondria are quite unique for its high ratio of fumarate reductase/succinate dehydrogenase activity (20.0), when compared with that of mammalian enzymes (0.05) [5]. The complex II was purified and was shown to be composed of four subunits with molecular masses of 68, 26, 15 and 13.5 kDa containing a flavin and a b-type cytochrome (cytochrome

Correspondence (present) address: A. Hata-Tanaka, Biochemical Research Laboratory, Morinaga Milk Industry Co., Ltd, 4-4-22 Meguro, Meguro-ku, Tokyo 153, Japan

Abbreviations: E_m , oxidation-reduction midpoint potential; $P_{1/2}$, apparent half-saturation parameter of ESR signal [1]

b-558) [5], as is well known for the mammalian complex II [6].

Succinate dehydrogenase in the mitochondrial complex II [6,7] and fumarate reductase in anaerobically growing bacteria [8] are two closely related enzyme systems. They have similar subunit compositions and active centers [FAD, iron-sulfur (Fe/S) clusters]. ESR studies have already suggested that both enzyme systems contain three corresponding types of Fe/S clusters, designated as clusters S-1, S-2 and S-3 [7,8]. Although compositions and ESR characteristics of these reaction centers are similar to each other, they play different roles in catalyzing the two directionally opposite enzyme reactions.

In this paper, Fe/S clusters from the Ascaris suum complex II were characterized by low-temperature ESR. The relation between the high fumarate reductase activity and the redox properties of these clusters was discussed by comparing them with the cases of bacterial fumarate reductase and mitochondrial succinate dehydrogenase.

2. MATERIALS AND METHODS

2.1. Preparation of Ascaris complex II

Mitochondria of Ascaris were prepared by the method of Takamiya et al. [2]. Complex II was solubilized in a solution containing 1% deoxycholate and 1% Triton X-100, and purified by fractionation by ammonium sulfate precipitation and by Sephadex G-200 column chromatography [3]. The purified complex II, thus obtained, was free from rhodo-quinone and other quinones [3]. Fumarate reductase activity (measured with methyl viologen as an electron donor [4]) of the purified complex II was 40 \(\mu\)min per mg protein.

2.2. ESR measurements

The purified complex II was suspended in $40 \mu l$ of 1 M Hepes/NaOH buffer (pH 7.0) and adjusted to a total volume of $200 \mu l$. The mixture was incubated with $4 \mu l$ of 1 M sodium succinate or saturated sodium dithionite solution for 20 min, transferred to an ESR sample tube under argon atmosphere at 20° C, and then frozen in liquid nitrogen. ESR signals were measured with an X-band Bruker ER 200D spectrometer (Bruker, FRG) equipped with a liquid helium model 900 cryostat (Oxford, England), as described previously [9,10].

3. RESULTS AND DISCUSSION

A strong ESR signal with positive and negative peaks at g = 2.025 and 2.014, respectively, was observed from the air-oxidized complex II at 13 K (fig.1a). Fig.1b shows a spectrum after reduction with 20 mM sodium succinate. It gives a positive peak at g = 2.033 in addition to the one seen in fig.1a and weaker signals at g = 1.939 and 1.920. Fig.1c shows a spectrum of the complex II reduced

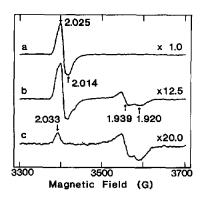


Fig.1. ESR signals of the Ascaris complex II. (a) Air-oxidized sample (5.3 mg protein/ml); (b) succinate-reduced sample (2.1 mg protein/ml); (c) dithionite-reduced sample (1.3 mg protein/ml). Samples contained the Ascaris complex II and 200 mM Hepes/NaOH buffer (pH 7.0). ESR measurement conditions were: modulation amplitude, 10 G; microwave power, 100 μW; sample temperature, 13 K.

by sodium dithionite. In this condition, the peaks at g = 2.025 and 2.014 disappeared and the signals at g = 2.033, 1.939 and 1.920 were detected.

The dependence on the microwave power of intensities of the negative peak at g = 1.920 was examined to study the spin relaxation rate of the signal at 13 K (fig.2). The apparent $P_{1/2}$ value of the signal at g = 1.920 of the succinate-reduced complex II was about 0.3 mW, while that from the dithionite-reduced sample was 30 mW. The difference of the $P_{1/2}$ value of the same signal under these conditions suggests that the complex has two interacting Fe/S clusters; one is reduced with succinate and shows microwave power saturation at 0.3 mW, and the other is reduced by dithionite only and enhances the spin relaxation rate of the former signals by spin-spin interactions which results in the higher $P_{1/2}$ value. These clusters can be assigned as S-1 and S-2, respectively [11]. The $E_{\rm m}$ values of clusters S-1 and S-2 are reported to be 0 and -260 mV, respectively, in bovine complex II [12], and -20 and -330 mV, respectively, in E. coli fumarate reductase [8]. In both cases, spinspin interactions between clusters S-1 and S-2 were reported; for example, the $P_{1/2}$ values of 0.07 mW (succinate-reduced, noninteracting condition) and 6.0 mW (dithionite-reduced, interacting condition) were reported in a mammalian mitochondrial succinate dehydrogenase preparation at 12 K [11,13]. The g values of the signals of the Ascaris cluster S-1 closely resemble those of the bovine mitochondrial centers (g = 2.03, 1.93 and 1.91) [6,12] and the E. coli fumarate reductase (g = 2.03, 1.93) and 1.91) [8]. The broad ESR signals of cluster S-2 showing strong rhombicity (g = 2.06, 1.99 and 1.85 for bovine [13] and g = 2.17, 1.90 and 1.68 for E. coli fumarate reductase [14]) could not be detected

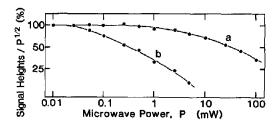


Fig.2. Microwave power saturation behavior of the ESR signal at g = 1.920 (a) under the dithionite-reduced condition; (b) under the succinate-reduced condition. Measurement conditions and sample preparations were the same as in fig.1.

in the Ascaris sample of this spin concentration (fig.1c).

The strong signal near g = 2.03 with positive and negative peaks observed in fig. la is due to cluster S-3 [15]. The $P_{1/2}$ value of the signal was 30 mW at 13 K and 0.4 mW at 7 K (not shown). The distinct dependence on temperature of the power saturation behavior reflects its short relaxation time. The signal of bovine S-3 also shows a short relaxation time (high $P_{1/2}$ value), and is diminished at temperatures higher than 15 K at a microwave power setting of 1 mW [15]. Thus, we measured the temperature dependence of the broad signal near g = 2.03 seen in fig.1b. At 50 K, the broad signal near g = 2.03 became small (fig.3), and typical peaks of cluster S-1 at g = 2.033, 1.939 and 1.920 were observed. Therefore, the higher field positive peak at g = 2.025 in fig.1b, which disappears at higher temperatures, could be attributed to the oxidized form of cluster S-3.

These results indicate that more than 15% of the g=2.025 signal remains after reduction with sodium succinate. This strongly suggests that S-3 in Ascaris is not fully reduced even by excess sodium succinate. However, S-3 in the bovine complex II was reported to be completely reduced under similar conditions (E_m value is 65 mV) [15]. The succinate-ubiquinone reductase activity of the Ascaris complex II used in this experiment is as high as that in the bovine complex II (2.0 μ g mol/min per mg protein) [16], suggesting its high reactivity with sodium succinate. Therefore, the E_m value of Ascaris S-3 may be lower than that of the sodium succinate/fumarate couple ($E_m = 30$ mV). The lower E_m value of cluster S-3 is a

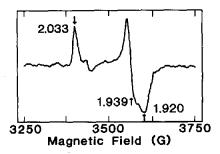


Fig. 3. ESR signals from succinate-reduced Ascaris complex II at 50 K. ESR measurements conditions were: modulation amplitude, 10 G; microwave power, 1.0 mW; protein concentration, 3.1 mg protein/ml.

common feature of fumarate reductase systems in Wolinella succinogenes ($E_{\rm m}=-24~{\rm mV}$) [17] and E. coli (-70 mV) [8]. We also reported that TTFA (2-thenoyltrifluoroacetone) shows no effect on the succinate dehydrogenase activity of the Ascaris complex II [16], although it is a potent inhibitor of the bovine complex II [18]. In the case of the bovine complex II, TTFA inhibits reoxidation of the cluster S-3 [19,20]. Thus, it indicates a difference in reactivity of the Ascaris cluster S-3 from the bovine mitochondrial cluster S-3. This may be related to the different behavior of the ESR signal of cluster S-3 observed here. These results indicate the unique characteristics of cluster S-3 in Ascaris, and help us to understand its high fumarate reductase activity.

Acknowledgements: We are grateful to Drs Y. Fujita, National Institute for Basic Biology (NIBB), Okazaki, and Y. Anraku, University of Tokyo, for their helpful advice and use of the laboratory facilities. We are also grateful to Mr H. Hattori and Mr Y. Kojima for help with ESR measurements performed at the Center for Chemical and Physical Analysis in NIBB. This study was carried out under the NIBB Cooperative Research Program (86-111).

REFERENCES

- Blum, H. and Ohnishi, T. (1980) Biochim. Biophys. Acta 621, 9-18.
- [2] Takamiya, S., Furushima, R. and Oya, H. (1984) Mol. Biochem. Parasitol. 13, 121-134.
- [3] Takamiya, S., Furushima, R. and Oya, H. (1986) Biochim. Biophys. Acta 848, 99-107.
- [4] Kita, K., Takamiya, S., Furushima, R., Ma, Y.-C. and Oya, H. (1988) Biochim. Biophys. Acta, in press.
- [5] Kita, K., Takamiya, S., Furushima, R., Ma, Y.-C. and Oya, H. (1987) Comp. Biochem. Physiol. 89, 31-34.
- [6] Hatefi, Y. (1985) Annu. Rev. Biochem. 54, 1015-1069.
- [7] Ohnishi, T. (1987) Curr. Top. Bioenerg. 15, 37-65.
- [8] Cole, S.T., Condon, C., Lemire, B.D. and Weiner, J.H. (1985) Biochim. Biophys. Acta 811, 381-403.
- [9] Hata, A., Kirino, Y., Matsuura, K., Itoh, S., Hiyama, T., Konishi, K., Kita, K. and Anraku, Y. (1985) Biochim. Biophys. Acta 810, 62-72.
- [10] Hata-Tanaka, A., Matsuura, K., Itoh, S. and Anraku, Y. (1987) Biochim. Biophys. Acta 893, 289-295.
- [11] Ohnishi, T. and Salerno, J.C. (1982) in: Iron-Sulfur Proteins (Spiro, T.G. ed.) vol.4, pp.285-327, John Wiley & Sons, New York.
- [12] Ohnishi, T., Salerno, J.C., Winter, D.B., Lim, J., Yu, C.A., Yu, L. and King, T.E. (1976) J. Biol. Chem. 251, 2094-2104.
- [13] Maguire, J.J., Johnson, M.K., Morningstar, J.E., Ackrell, B.A.C. and Kearney, E.B. (1985) J. Biol. Chem. 260, 10909-10912.

- [14] Cammack, R., Patil, D.S. and Weiner, J.H. (1986) Biochim. Biophys. Acta 870, 545-551.
- [15] Ohnishi, T., Lim, J., Winter, D.B. and King, T.E. (1976) J. Biol. Chem. 251, 2105-2109.
- [16] Ma, Y.-C., Kita, K., Hamajima, F. and Oya, H. (1987) Jap. J. Parasitol. 36, 107-117.
- [17] Unden, G., Albracht, S.P.J. and Kröger, A. (1984) Biochim. Biophys. Acta 767, 460-469.
- [18] Mowery, P.C., Ackrell, B.A.C., Singer, T.P., White, G.A. and Thorn, G.D. (1976) Biochem. Biophys. Res. Commun. 71, 354-361.
- [19] Mowery, P.C., Steenkamp, D.J., Ackrell, B.A.C., Singer, T.P. and White, G.A. (1977) Arch. Biochem. Biophys. 178, 495-506.
- [20] Salerno, J.C. and Ohnishi, T. (1980) Biochem. J. 192, 769-781.